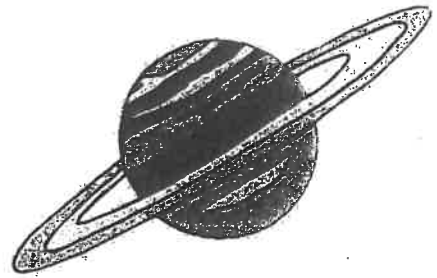
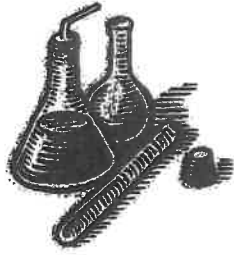


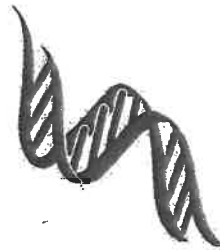
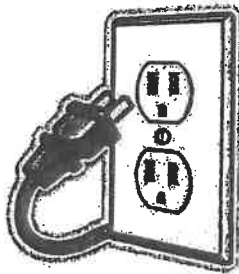
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Data



Booklet



UNITS AND ABBREVIATIONS

Quantity	Unit	Symbol
Distance (d)	metre	m
Voltage (V)	volt	V
Current (I)	ampere	A
Resistance (R)	ohm	Ω
Power (P)	watt	W
Work (W)	joule	J
Force (F)	newton	N
Energy (E)	joule	J
Time (t)	second	s
Time (t)	minute	min
Time (t)	hour	h
Time (t)	year	a

FORMULAE

$V = IR$	$P = VI$	$E = Pt$
$R = \frac{V}{I}$	$I = \frac{P}{V}$	$P = \frac{E}{t}$
$I = \frac{V}{R}$	$V = \frac{P}{I}$	$t = \frac{E}{P}$

ALPHABETICAL LISTING OF THE ELEMENTS

Element	Symbol	Atomic Number	Element	Symbol	Atomic Number
Actinium	Ac	89	Mendelevium	Md	101
Aluminium	Al	13	Mercury	Hg	80
Americium	Am	95	Molybdenum	Mo	42
Antimony	Sb	51	Neodymium	Nd	60
Argon	Ar	18	Neon	Ne	10
Arsenic	As	33	Neptunium	Np	93
Astatine	At	85	Nickel	Ni	28
Barium	Ba	56	Niobium	Nb	41
Berkellium	Bk	97	Nitrogen	N	7
Beryllium	Be	4	Nobelium	No	102
Bismuth	Bi	83	Osmium	Os	76
Bohrium	Bh	107	Oxygen	O	8
Boron	B	5	Palladium	Pd	46
Bromine	Br	35	Phosphorus	P	15
Cadmium	Cd	48	Platinum	Pt	78
Calcium	Ca	20	Plutonium	Pu	94
Californium	Cf	98	Polonium	Po	84
Carbon	C	6	Potassium	K	19
Cerium	Ce	58	Praseodymium	Pr	59
Cesium	Cs	55	Promethium	Pm	61
Chlorine	Cl	17	Protactinium	Pa	91
Chromium	Cr	24	Radium	Ra	88
Cobalt	Co	27	Radon	Rn	86
Copper	Cu	29	Rhenium	Re	75
Curium	Cm	96	Rhodium	Rh	45
Darmstadtium	Ds	110	Roentgenium	Rg	111
Dubnium	Db	105	Rubidium	Rb	37
Dysprosium	Dy	66	Ruthenium	Ru	44
Einsteinium	Es	99	Rutherfordium	Rf	104
Erbium	Er	68	Samarium	Sm	62
Europium	Eu	63	Scandium	Sc	21
Fermium	Fm	100	Seaborgium	Sg	106
Fluorine	F	9	Selenium	Se	34
Francium	Fr	87	Silicon	Si	14
Gadolinium	Gd	64	Silver	Ag	47
Gallium	Ga	31	Sodium	Na	11
Germanium	Ge	32	Strontium	Sr	38
Gold	Au	79	Sulfur	S	16
Hafnium	Hf	72	Tantalum	Ta	73
Hassium	Hs	108	Technetium	Tc	43
Helium	He	2	Tellurium	Te	52
Holmium	Ho	67	Terbium	Tb	65
Hydrogen	H	1	Thallium	Tl	81
Indium	In	49	Thorium	Th	90
Iodine	I	53	Thulium	Tm	69
Iridium	Ir	77	Tin	Sn	50
Iron	Fe	26	Titanium	Ti	22
Krypton	Kr	36	Tungsten	W	74
Lanthanum	La	57	Uranium	U	92
Lawrencium	Lr	103	Venadium	V	23
Lead	Pb	82	Xenon	Xe	54
Lithium	Li	3	Ytterbium	Yb	70
Lutetium	Lu	71	Yttrium	Y	39
Magnesium	Mg	12	Zinc	Zn	30
Manganese	Mn	25	Zirconium	Zr	40
Mendelevium	Md	101			
Mercury	Hg	80			
Molybdenum	Mo	42			
Neodymium	Nd	60			
Neon	Ne	10			
Neptunium	Np	93			
Nickel	Ni	28			
Niobium	Nb	41			
Nitrogen	N	7			
Nobelium	No	102			
Osmium	Os	76			
Oxygen	O	8			
Palladium	Pd	46			
Phosphorus	P	15			
Platinum	Pt	78			
Plutonium	Pu	94			
Polonium	Po	84			
Potassium	K	19			
Praseodymium	Pr	59			
Promethium	Pm	61			
Protactinium	Pa	91			
Radium	Ra	88			
Radon	Rn	86			
Rhenium	Re	75			
Rhodium	Rh	45			
Roentgenium	Rg	111			
Rubidium	Rb	37			
Ruthenium	Ru	44			
Rutherfordium	Rf	104			
Samarium	Sm	62			
Scandium	Sc	21			
Seaborgium	Sg	106			
Selenium	Se	34			
Silicon	Si	14			
Silver	Ag	47			
Sodium	Na	11			
Strontium	Sr	38			
Sulfur	S	16			
Tantalum	Ta	73			
Technetium	Tc	43			
Tellurium	Te	52			
Terbium	Tb	65			
Thallium	Tl	81			
Thorium	Th	90			
Thulium	Tm	69			
Tin	Sn	50			
Titanium	Ti	22			
Tungsten	W	74			
Uranium	U	92			
Venadium	V	23			
Xenon	Xe	54			
Ytterbium	Yb	70			
Yttrium	Y	39			
Zinc	Zn	30			
Zirconium	Zr	40			

PERIODIC TABLE OF THE ELEMENTS

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
1 H Hydrogen 1.0																	2 He Helium 4.0	
3 Li Lithium 6.9	4 Be Beryllium 9.0													7 N Nitrogen 14.0	8 O Oxygen 16.0	9 F Fluorine 19.0	10 Ne Neon 20.2	
11 Na Sodium 23.0	12 Mg Magnesium 24.3												14 Si Silicon 28.1	15 P Phosphorus 31.0	16 S Sulphur 32.1	17 Cl Chlorine 35.5	18 Ar Argon 39.9	
19 K Potassium 39.1	20 Ca Calcium 40.1	21 Sc Scandium 45.0	22 Ti Titanium 47.9	23 V Vanadium 50.9	24 Cr Chromium 52.0	25 Mn Manganese 54.9	26 Fe Iron 55.8	27 Co Cobalt 58.9	28 Ni Nickel 58.7	29 Cu Copper 63.5	30 Zn Zinc 65.4	31 Ga Gallium 69.7	32 Ge Germanium 72.6	33 As Arsenic 74.9	34 Se Selenium 79.0	35 Br Bromine 79.9	36 Kr Krypton 83.8	
37 Rb Rubidium 85.5	38 Sr Strontium 87.6	39 Y Yttrium 88.9	40 Zr Zirconium 91.2	41 Nb Niobium 92.9	42 Mo Molybdenum 95.9	43 Tc Technetium (98)	44 Ru Ruthenium 101.1	45 Rh Rhodium 102.9	46 Pd Palladium 106.4	47 Ag Silver 107.9	48 Cd Cadmium 112.4	49 In Indium 114.8	50 Sn Tin 118.7	51 Sb Antimony 121.8	52 Te Tellurium 127.6	53 I Iodine 126.9	54 Xe Xenon 131.3	
55 Cs Cesium 132.9	56 Ba Barium 137.3	57 La Lanthanum 138.9	72 Hf Hafnium 178.5	73 Ta Tantalum 180.9	74 W Tungsten 183.8	75 Re Rhenium 186.2	76 Os Osmium 190.2	77 Ir Iridium 192.2	78 Pt Platinum 195.1	79 Au Gold 197.0	80 Hg Mercury 200.6	81 Tl Thallium 204.4	82 Pb Lead 207.2	83 Bi Bismuth 209.0	84 Po Polonium (209)	85 At Astatine (210)	86 Rn Radon (222)	
87 Fr Francium (223)	88 Ra Radium (226)	89 Ac Actinium (227)	104 Rf Rutherfordium (261)	105 Db Dubnium (262)	106 Sg Seaborgium (263)	107 Bh Bohrium (262)	108 Hs Hassium (265)	109 Mt Meitnerium (266)										

14
Si
Silicon
28.1

Atomic Number
Symbol
Name
Atomic Mass

58 Ce Cerium 140.1	59 Pr Praseodymium 140.9	60 Nd Neodymium 144.2	61 Pm Promethium (145)	62 Sm Samarium 150.4	63 Eu Europium 152.0	64 Gd Gadolinium 157.3	65 Tb Terbium 158.9	66 Dy Dysprosium 162.5	67 Ho Holmium 164.9	68 Er Erbium 167.3	69 Tm Thulium 168.9	70 Yb Ytterbium 173.0	71 Lu Lutetium 175.0
90 Th Thorium 232.0	91 Pa Protactinium 231.0	92 U Uranium 238.0	93 Np Neptunium (237)	94 Pu Plutonium (244)	95 Am Americium (243)	96 Cm Curium (247)	97 Bk Berkelium (247)	98 Cf Californium (251)	99 Es Einsteinium (252)	100 Fm Fermium (257)	101 Md Mendelevium (258)	102 No Nobelium (259)	103 Lr Lawrencium (262)

Based on mass of C¹² at 12.00.

Values in parentheses are the masses of the most stable or best known isotopes for elements which do not occur naturally.

PERIODIC TABLE OF THE ELEMENTS

1 H Hydrogen 1.0	NON-METALS																2 He Helium 4.0		
3 Li Lithium 6.9	4 Be Beryllium 9.0																	9 F Fluorine 19.0	10 Ne Neon 20.2
11 Na Sodium 23.0	12 Mg Magnesium 24.3																	17 Cl Chlorine 35.5	18 Ar Argon 39.9
19 K Potassium 39.1	20 Ca Calcium 40.1	21 Sc Scandium 45.0	22 Ti Titanium 47.9	23 V Vanadium 50.9	24 Cr Chromium 52.0	25 Mn Manganese 54.9	26 Fe Iron 55.8	27 Co Cobalt 58.9	28 Ni Nickel 58.7	29 Cu Copper 63.5	30 Zn Zinc 65.4	31 Ga Gallium 69.7	32 Ge Germanium 72.6	33 As Arsenic 74.9	34 Se Selenium 78.0	35 Br Bromine 79.9	36 Kr Krypton 83.8		
37 Rb Rubidium 85.5	38 Sr Strontium 87.6	39 Y Yttrium 88.9	40 Zr Zirconium 91.2	41 Nb Niobium 92.9	42 Mo Molybdenum 95.9	43 Tc Technetium (98)	44 Ru Ruthenium 101.1	45 Rh Rhodium 102.9	46 Pd Palladium 106.4	47 Ag Silver 107.9	48 Cd Cadmium 112.4	49 In Indium 114.8	50 Sn Tin 118.7	51 Sb Antimony 121.8	52 Te Tellurium 127.6	53 I Iodine 126.9	54 Xe Xenon 131.3		
55 Cs Cesium 132.9	56 Ba Barium 137.3	57 La Lanthanum 138.9	58 Ce Cerium 140.1	59 Pr Praseodymium 140.9	60 Nd Neodymium 144.2	61 Pm Promethium (145)	62 Sm Samarium 150.4	63 Eu Europium 152.0	64 Gd Gadolinium 157.3	65 Tb Terbium 158.9	66 Dy Dysprosium 162.5	67 Ho Holmium 164.9	68 Er Erbium 167.3	69 Tm Thulium 168.9	70 Yb Ytterbium 173.0	71 Lu Lutetium 175.0			
87 Fr Francium (223)	88 Ra Radium (226)	89 Ac Actinium (227)	90 Th Thorium 232.0	91 Pa Protactinium 231.0	92 U Uranium 238.0	93 Np Neptunium (237)	94 Pu Plutonium (244)	95 Am Americium (243)	96 Cm Curium (247)	97 Bk Berkelium (247)	98 Cf Californium (251)	99 Es Einsteinium (252)	100 Fm Fermium (257)	101 Md Mendelevium (258)	102 No Nobelium (259)	103 Lr Lawrencium (262)			
Alkali Metals																Noble Gases			
Alkaline Earth Metals																Halogens			

Atomic Number	22	44	34	Ion charge(s)
Symbol	Ti	Ti	Ti	
Name	Titanium	Titanium	Titanium	
Atomic Mass	47.9	47.9	47.9	

Based on mass of C-12 at 12.00.

Any value in parentheses is the mass of the most stable or best known isotope for elements which do not occur naturally.

NAMES, FORMULAE AND CHARGES OF SOME COMMON IONS

Positive Ions

Al^{3+} Aluminium	Li^+ Lithium
NH_4^+ Ammonium	Mg^{2+} Magnesium
Ba^{2+} Barium	Mn^{2+} Manganese(II)
Ca^{2+} Calcium	Mn^{4+} Manganese(IV)
Cr^{2+} Chromium(II)	Hg_2^{2+} Mercury(I)
Cr^{3+} Chromium(III)	Hg^{2+} Mercury(II)
Cu^+ Copper(I)	K^+ Potassium
Cu^{2+} Copper(II)	Ag^+ Silver
H^+ Hydrogen	Na^+ Sodium
Fe^{2+} Iron(II)	Sn^{2+} Tin(II)
Fe^{3+} Iron(III)	Sn^{4+} Tin(IV)
Pb^{2+} Lead(II)	Zn^{2+} Zinc
Pb^{4+} Lead(IV)	

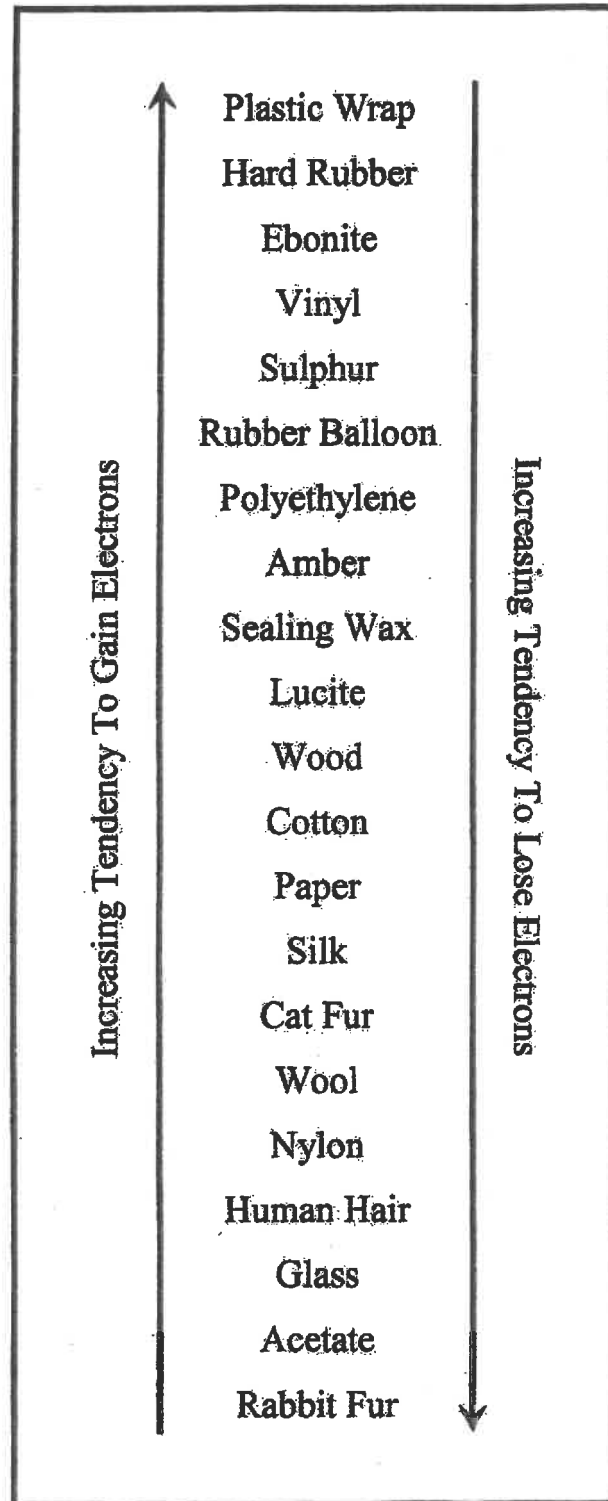
Negative Ions

CH_3COO^- Acetate	OH^- Hydroxide
Br^- Bromide	ClO^- Hypochlorite
CO_3^{2-} Carbonate	I^- Iodide
ClO_3^- Chlorate	NO_3^- Nitrate
Cl^- Chloride	NO_2^- Nitrite
ClO_2^- Chlorite	O^{2-} Oxide
CrO_4^{2-} Chromate	ClO_4^- Perchlorate
CN^- Cyanide	MnO_4^- Permanganate
$Cr_2O_7^{2-}$ Dichromate	PO_4^{3-} Phosphate
F^- Fluoride	PO_3^{3-} Phosphite
HCO_3^- Hydrogen carbonate, bicarbonate	SO_4^{2-} Sulphate
HSO_4^- Hydrogen sulphate, bisulphate	S^{2-} Sulphide
HS^- Hydrogen sulphide, bisulphide	SO_3^{2-} Sulphite
HSO_3^- Hydrogen sulphite, bisulphite	

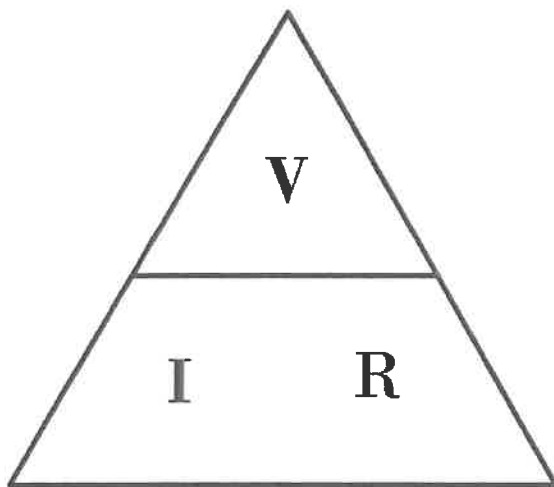
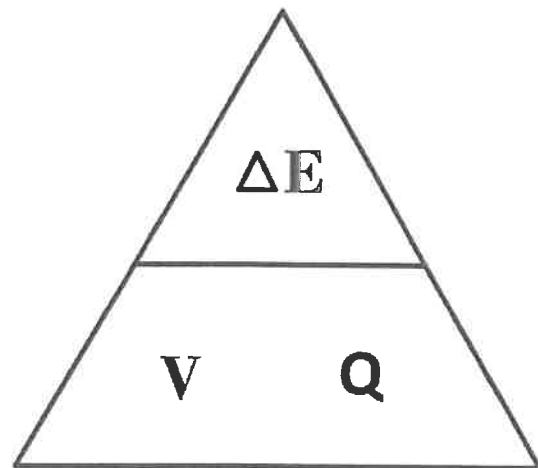
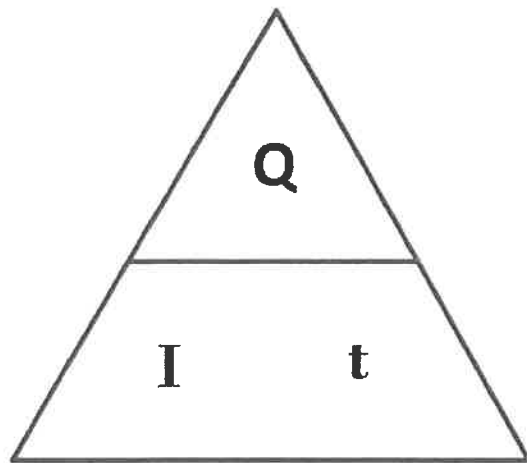
Prefixes:

1	Mono	6	Hexa
2	Di	7	Hepta
3	Tri	8	Octa
4	Tetra	9	Nona
5	Penta	10	Deca

THE ELECTROSTATIC SERIES



Circuit Formulas



$$1.0 \text{ C} = 6.25 \times 10^{18} \text{ e-}$$

